

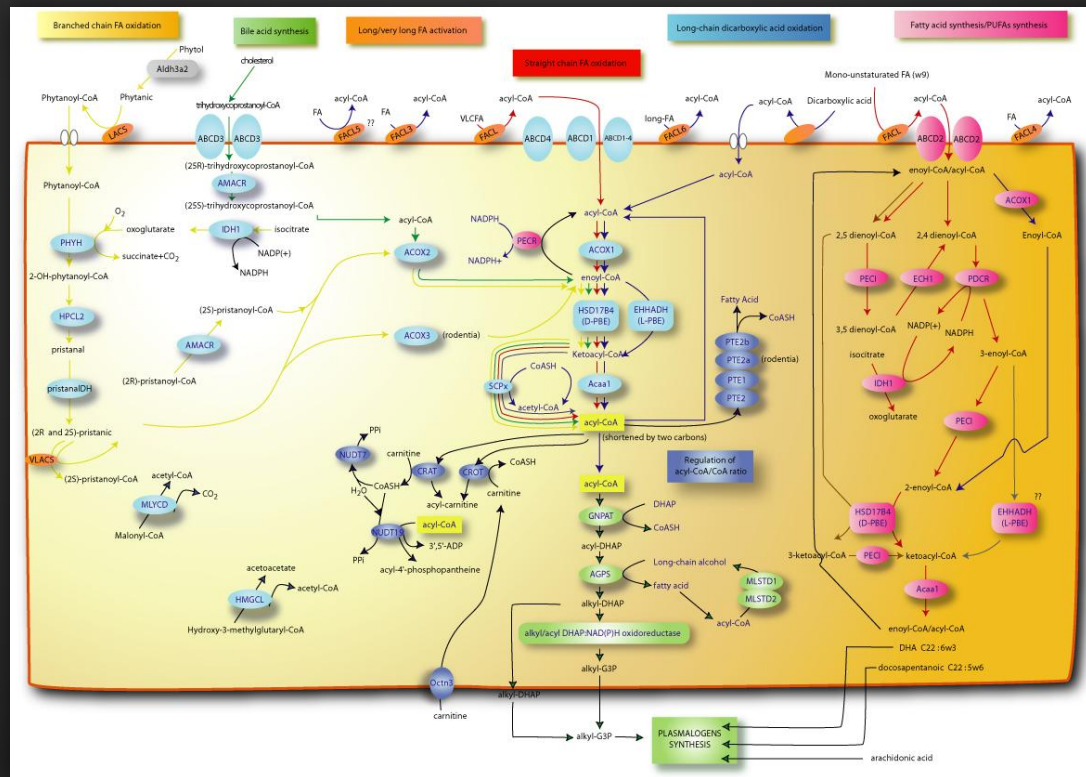
Synthetic Biological Systems

1. Metabolic Engineering with Synthetic Biology

What is Metabolic Engineering?

"Metabolic engineering is the practice of optimizing genetic and regulatory processes within cells to increase the cells' production of a certain substance"

Wikipedia



Metabolic engineering for biosynthesis

An attractive alternative to chemical synthesis

- Generate fuels from renewable resources
- Convert biomass into chemicals – both bulk and speciality
- Produce therapeutic compounds that are a chemical challenge
- Tap into nature's huge diversity

A 21st century goldmine

- Major push by US Department of Energy – “12 value-added compounds”
- Huge investments from BP, Chevron, Bill Gates
- Green alternative to the petrochemical industry



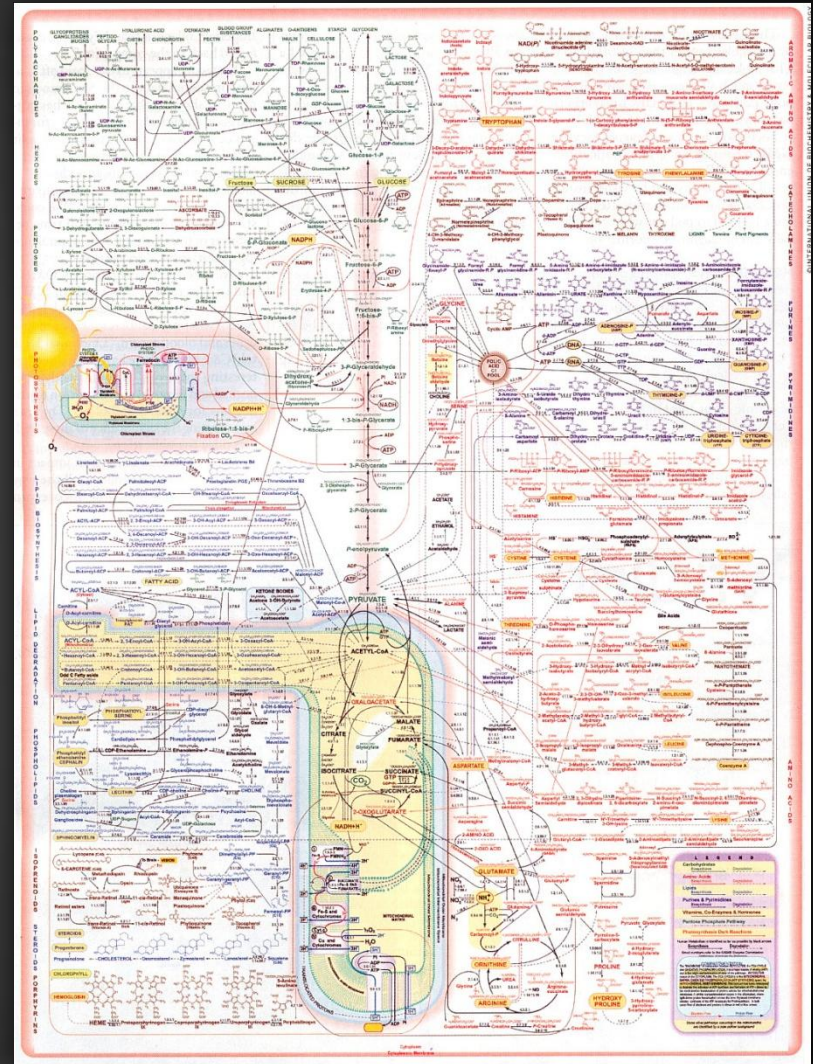
Lecture Content

In this lecture we'll learn about:

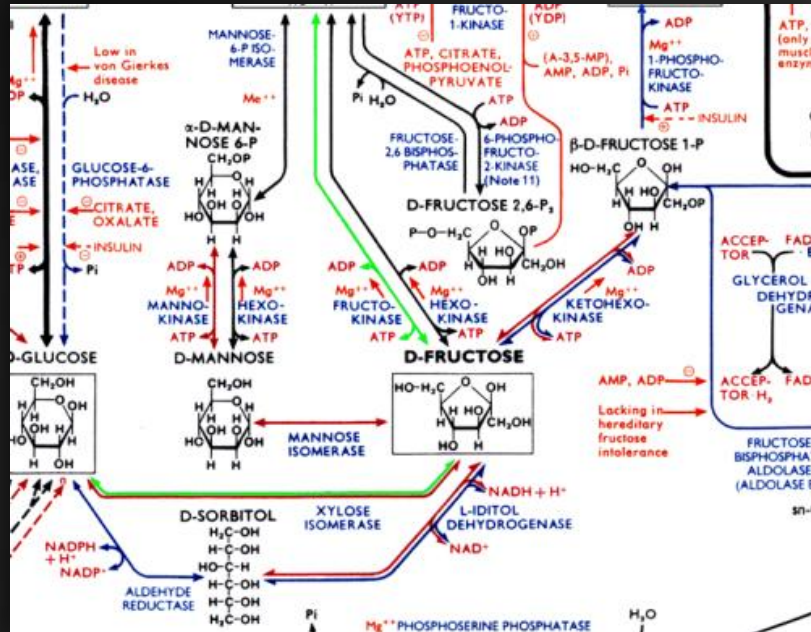
1. The complexity of metabolism in a cell and the major challenges in metabolic engineering
2. What synthetic biology offers for metabolic engineering
3. The data and models associated with metabolic engineering
4. A major project to engineer the synthesis of a crucial anti-malarial in both *E.coli* and yeast cells
5. Engineering bacteria to produce biopetrol from sunlight and CO₂
6. Engineering bacteria to turn plant waste into diesel biofuels
7. Novel synthetic biology methods to optimise metabolic pathways
8. The future for synthetic biology and metabolic engineering

Cellular metabolism is a complex network

- Cellular metabolism is a complex inter-dependent network
- The metabolome is all the small molecules of a cell
- Metabolic networks are defined by pathways
- Flux is the rate of turnover of molecules through a pathway
- Flux is regulated by the enzymes in a pathway



Metabolic engineering is like managing traffic



- Carbons and other atoms are the people
- Metabolites are their location
- Enzymes are the roads and railways they travel on

Synthetic biology for metabolic engineering

Synthetic Biology offers:

1. Predictable, designed genetic engineering
2. Regulation of gene expression
3. Insertion of new genes and new functions
4. Scalability – the addition or modification of many genes

This gives metabolic engineering:

1. Engineering by design
2. Control of enzyme levels, and so control of flux
3. Synthesis of new products
4. Assembly of whole new pathways

Basic metabolic engineering: change gene expression

- Increase the expression of enzymes involved in synthesis
 - add extra copies of enzyme gene into the cell
 - over-express gene using strong, regulated promoter
- Remove (knock-out) enzymes that takeaway from synthesis
 - use homologous recombination to delete host genes
- Careful balancing act – toxic precursors, growth rates
 - accumulation of products inhibits pathway production
 - many intermediates are toxic at high levels
 - cell global metabolism is always effected
 - resources must be taken from somewhere
 - slower growth is less yield

Further metabolic engineering: change the genes

- Modify the enzymes

remove regulatory sub-units, change cell localisation

- Add enzymes from across nature – heterologous pathways

400,000 known proteins and millions unknown

- Design new enzymes from scratch

de novo design of active sites

- Directed evolution

mutate the enzymes, select improvements, repeat

Measuring the results of metabolic engineering

- GC Mass Spec
 - Always on CSIGas liquid chromatography plus mass spectrometry
- Growth rate
- Radio-labelled atoms
- Enzyme assays
- Gene expression data
 - mRNA levels: cDNA array chips / sequencing
 - protein levels: proteomics / MALDI-TOF



Predicting metabolic engineering

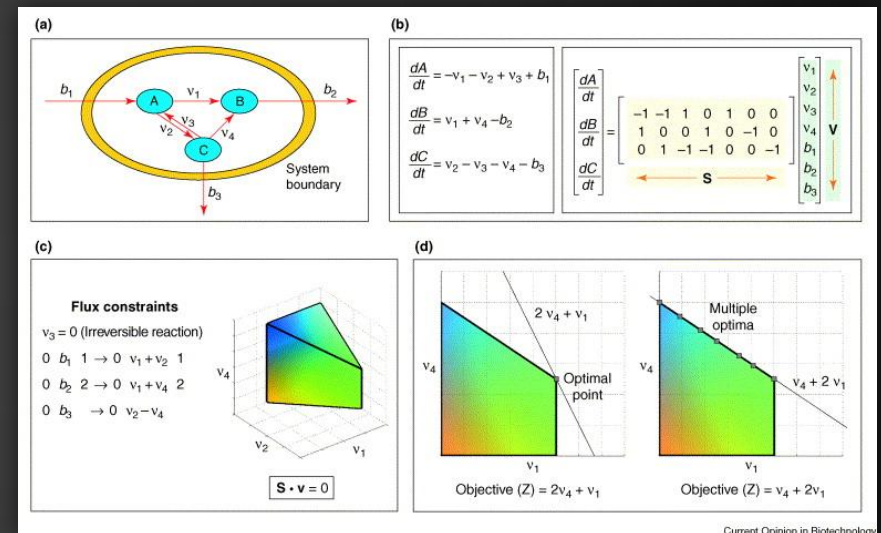
How to avoid bottlenecks, toxins, and negative feedback?

- Flux-balance analysis (FBA)

Mathematical analysis of the metabolic network under perturbations

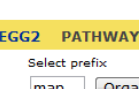
Model built using GC-MS data under different conditions

Most metabolism across the cell relies on the levels of a few chemicals:
ATP/ADP, NAD⁺/NADH, NADP⁺/NADPH, Acyl-CoAs, TCA cycle



Next Gen: FBA with gene expression = Metabolomics + Systems Biology

Bioinformatics for Metabolic Engineering



KEGG PATHWAY Database

Wiring diagrams of molecular interactions, reactions,

KEGG2	PATHWAY	BRITE	MODULE	DISEASE	DRUG	GENES	GENOM
-------	---------	-------	--------	---------	------	-------	-------

Select prefix

Enter keywords

Pathway Maps

KEGG PATHWAY is a collection of manually drawn pathway maps (see [new maps](#) and [last updates](#)) representing our knowledge on the molecular interaction and for:

0. Global Map
1. Metabolism
 - Carbohydrate
 - Energy
 - Lipid
 - Nucleotide
 - Amino acid
 - Other amino acid
 - Cofactor/vitamin
 - Terpenoid/PK
 - Other secondary metabolite
 - Xenobiotics
2. Genetic Information Processing
3. Environmental Information Processing
4. Cellular Processes
5. Organismal Systems
6. Human Diseases

and also on the structure relationships (KEGG drug structure maps) in:

7. Drug Development

Pathway Mapping

KEGG PATHWAY mapping is the process to map molecular datasets, especially large-scale genomics, transcriptomics, proteomics, and metabolomics, to the KEGG pathway database for the interpretation of higher-level systemic functions.

- Search objects in KEGG pathways
- Color objects in KEGG pathways

0. Global Map

0.1 Metabolism

Metabolic pathways [zoom out]	Launch KEGG Atlas
Biosynthesis of secondary metabolites [zoom out]	Launch KEGG Atlas
Microbial metabolism in diverse environments [zoom out] <i>New!</i>	Launch KEGG Atlas

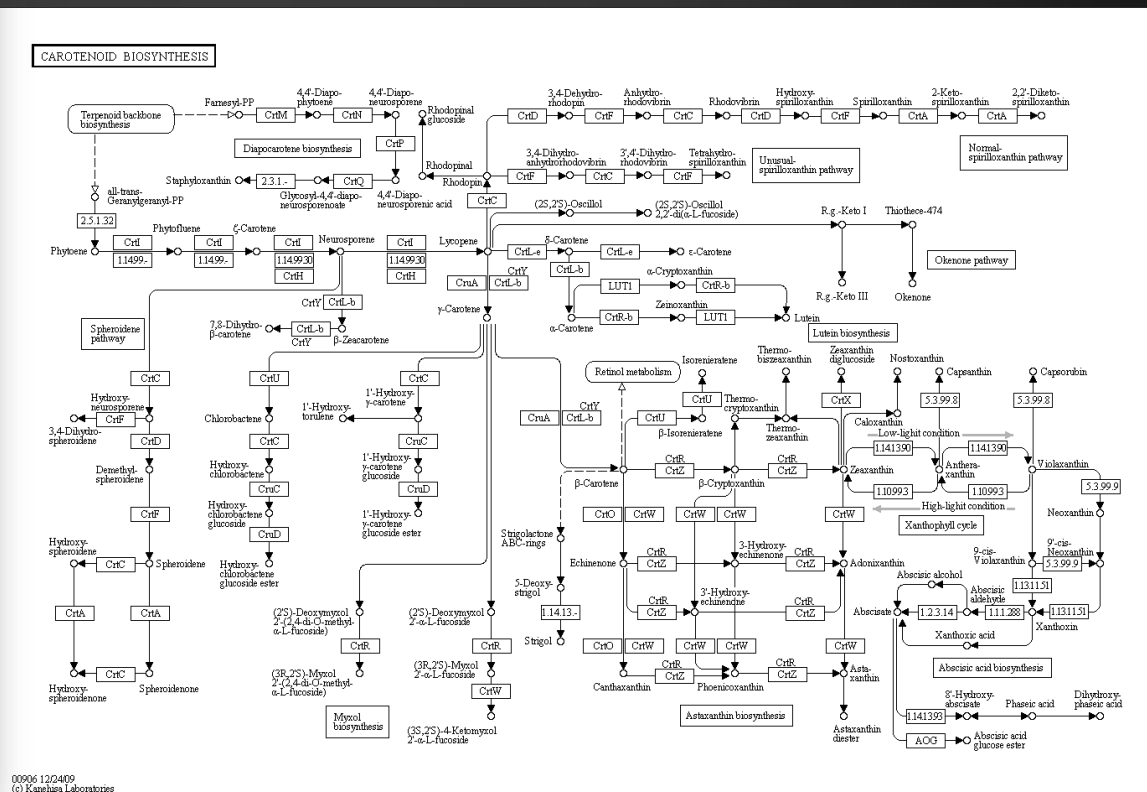
1. Metabolism

1.1 Carbohydrate Metabolism

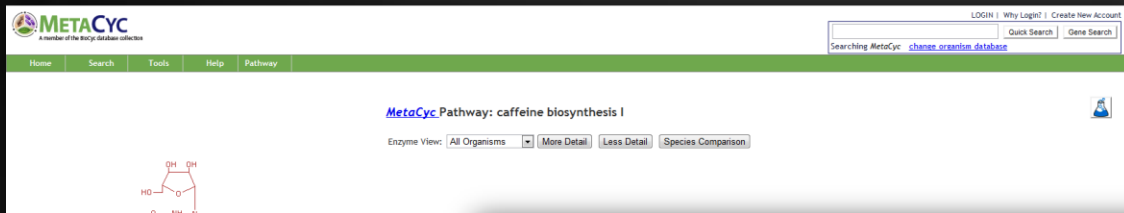
Glycolysis / Gluconeogenesis	Enzymes
Citrate cycle (TCA cycle)	Compounds with

Kegg Pathway has thousands of annotated pathways

Example: Carotenoids – e.g. Lycopene, Carotene



Bioinformatics for Metabolic Engineering



MetaCyc

Example: Caffeine synthesis

This view shows enzymes only for those organisms listed below

If an enzyme name is shown in bold, there is experimental evidence for this enzymatic activity.

Superclasses: Biosynthesis -> Secondary Metabolites Biosynthesis -> Nitrogen-Containing Secondary Compounds Biosynthesis -> Alkaloids Biosynthesis -> Purine Alkaloids Biosynthesis -> Caffeine Biosynthesis

Some taxa known to possess this pathway include [?](#) : [Camellia sinensis](#) , [Camellia sinensis assamica](#) , [Camellia taliensis](#) , [Coffea arabica](#) , [Coffea canephora](#)

Expected Taxonomic Range: [Malvaceae](#) , [Rubiaceae](#) , [Theaceae](#)

Summary:

Caffeine, 1,3,7-trimethylxanthine, is one of the best known purine alkaloids and also the most abundant one in nature, followed by theobromine, the immediate precursor of caffeine (3,7-dimethylxanthine) ([Schubert et al.](#)). Caffeine presents in high concentrations in many coffee plants and tea plants. It can accumulate up to 25 dry weight in seeds and young expanding leaves of *Coffea arabica*, and up to 35 dry weight in young leaves of tea (*Camellia sinensis*). The exact physiological role of endogenous purine alkaloids in plants is undetermined. Their hypothetical roles include being a defense chemical where high concentrations of caffeine accumulated in young leaves, fruits and flower buds help protect these soft tissues from predators, or being an autotoxic chemical where caffeine released from seed coats into the soil inhibits germination of other seeds. However there is little evidence supports the hypothesis.

The depicted pathway herein illustrates the major route of caffeine biosynthesis, from xanthosine. It involves three SAM-dependent N-methyltransferase activities, namely xanthosine N-methyltransferase, 7-methylxanthosine N-methyltransferase, and theobromine N-methyltransferase. 7-methylxanthosine synthesis of coffee was shown have the specific xanthosine N-methyltransferase activity. It specifically converts xanthosine to 7-methylxanthosine (the first methylation step). XMP is not an effective substrate. Theobromine synthase has also been characterized from coffee which has the specific 7-methylxanthosine N-methyltransferase activity. It converts 7-methylxanthosine to theobromine (the second methylation step). Caffeine synthesis has been characterized from both coffee and tea. Caffeine synthase dual function from coffee has been characterized with 7-methylxanthosine N-methyltransferase and theobromine to caffeine, the second and the third methylation steps. In fruits are the major sites of caffeine biosynthesis in coffee plants, where high levels of transcripts of genes encoding the three methyltransferase activities are also found [Jefu03].

Free purine nucleotides are the major resources of xanthosine. Xanthosine is derived from salvage of adenine and guanine nucleotides (see [purine nucleosides salvage II \(plant\)](#)). Newly synthesized IMP from de novo purine biosynthesis, and adenosine released from the SAM cycle (see [S-adenosyl-L-methionine cycle II](#)) may also be converted to xanthosine and enter the caffeine biosynthesis pathway.

In addition to the major route, caffeine may also be synthesized via a few minor routes such as from 7-methylxanthine via paraxanthine (see [caffeine biosynthesis II \(via paraxanthine\)](#).)

Citations: [[Misako04](#), [Ogawa01](#), [Mizuno03](#), [Mizuno03a](#), [Kato96](#)]

Variants: [caffeine biosynthesis II \(via paraxanthine\)](#)

References

Ashihara04: Ashihara H, Suzuki T (2004). "Distribution and biosynthesis of caffeine in plants." *Front Biosci* 9:1864-76. PMID: 14977593

Kato96: Kato, Misako, Kanehara, Tomomi, Shimizu, Hisayo, Suzuki, Takeo, Gillies, Fiona, Ashihara, Hiroshi (1996). "Caffeine biosynthesis in young leaves of *Camellia sinensis*: In vitro studies on N-methyl transferase activity involved in the conversion of xanthosine to caffeine." *Physiologia Plantarum* 98:629-36.

Misako04: Misako K, Kouichi M (2004), "Caffeine synthase and related methyltransferases in plants," *Front Biosci* 9:1833-42, PMID: 14977590

Mizuno03: Mizuno K, Kato M, Ichino F, Yoneyama N, Fujimura T, Ashihara H (2003). "The first committed step reaction of caffeine biosynthesis: 7-methylxanthosine synthase is closely homologous to caffeine synthases in coffee (*Coffea arabica* L.)." *FEBS Lett* 547(1-3):56-60. PMID: 12860386

Mizuno03a: Mizuno K, Okuda A, Kato M, Yoneyama N, Tanaka H, Ashihara H, Fujimura T (2003). "Isolation of a new dual-functional caffeine synthase gene encoding an enzyme for the conversion of 7-methylxanthine to caffeine from coffee (*Coffea arabica* L.)." *FEBS Lett* 534(1-3):75-81. PMID: 12527364

Ogawa01: Ogawa M, Heral Y, Kolzumi N, Kusano T, Sano H (2001). "7-Methylxanthine methyltransferase of coffee plants. Gene isolation and enzymatic properties." *J Biol Chem* 276(11):8213-8. PMID: 11108716

Uefuji03: Uefuji H, Ogita S, Yamaguchi Y, Koizumi N, Sano H (2003). "Molecular cloning and functional characterization of three distinct N-methyltransferases involved in the caffeine biosynthetic pathway in coffee plants." *Plant Physiol* 132(1):372-80. PMID: 12746542

References Related to Enzymes, Genes, Subpathways, and Substrates of this Pathway

Forsythe97: Forsythe RG, Karp PD, Mavrouniotis ML (1997). "Estimation of Equilibrium Constants Using Automated Group Contribution Methods." *CABIOS* 13(5):537-543.

Kato00: Kato M, Mizuno K, Crozier A, Fujimura T, Ashihara H (2000). "Caffeine synthase gene from tea leaves." *Nature* 406(6799):956-7. PMID: 10984041

Kato99: Kato M, Mizuno K, Fujimura T, Iwama M, Irie M, Crozier A, Ashihara H (1999). "Purification and characterization of caffeine synthase from tea leaves." *Plant Physiol* 120(2):579-86. PMID: 10364410

Negishi88: Negishi, Osamu, Ozawa, Tetsuo, Imagawa, Hiroshi (1988). "N-methyl nucleosidase from tea leaves." *Agric. Biol. Chem.* 52(1):169-175.

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Please cite the following article in publications resulting from the use of MetaCyc: [Caspi et al, Nucleic Acids Research 38:D473-D479 2010](#)

Page generated by SRI International [Pathway Tools](#) version 14.5 on Thu Jan 20, 2011, Biocyc12.

Bioinformatics for Metabolic Engineering

- BRENDA – gives hard information on enzymes
- For metabolic engineering it links to MetaCyc and Kegg



EC-Number Enzyme Name Organism Protein Full text Advanced Search

Search Display 10 entries

New BRENDA release online since January, 5th 2011
New publications on BRENDA

Nomenclature	Reaction & Specificity	Functional Parameters
Enzyme Names EC Number Common/ Recommended Name Systematic Name Synonyms CAS Registry Number	Pathway Catalysed Reaction Reaction Type Natural Substrates and Products Substrates and Products Substrates Natural Substrate Products Natural Product Inhibitors Cofactors Metals/Ions Activating Compounds Ligands Ligand Views Biochemicals Reactions Aligned	Km Value kcat/Km Value NEW Ki Value IC50 Value pI Value Turnover Number Specific Activity pH Optimum pH Range Temperature Optimum Temperature Range
Isolation & Preparation		Organism-related information
Purification Cloned Expression NEW Renatured Crystallization		Organism Source Tissue Localization Protein-Specific Search
Stability	Enzyme Structure	Disease & References
pH Stability Temperature Stability General Stability Organic Solvent Stability Oxidation Stability Storage Stability	Sequence/ SwissProt link 3D-Structure/ PDB link Molecular Weight Subunits Posttranslational Modification	Application & Engineering
		Engineering Application

Webmaster: Maurice Scheer

Enzyme Nomenclature
EC number
Recommended Name
Reaction
Reaction Type
Pathway
Systematic Name
Synonyms
CAS Registry Number
Enzyme-Ligand Interactions
Substrate/Product
Natural Substrates
Cofactor
Metals and Ions
Inhibitors
Activating Compound
Functional Parameters
KM Value
Turnover Number
kcat/KM Value
Ki Value
IC50 Value
Specific Activity
pH Optimum
pH Range
Temperature Optimum
Temperature Range
pl Value
Organism related information
Source Tissue
Localization
Organism
General Information
Enzyme Structure
AA Sequence

EC 3.5.1.11 - penicillin amidase

fluoride	coli		171892	image
phenylmethylsulfonyl fluoride	Escherichia coli	reactivation by incubation with 6-aminopenicillanic acid or proteins from E. coli	171894	2D-image
[bmim]dca	Escherichia coli	the enzyme activity in 25% [bmim]dca is 1.6fold than activity in water. The enzyme activity is decreased in higher concentration of [bmim]dca	685630	2D-image
Mn2+	Bacillus megaterium		680527	2D-image
additional information	Bacillus megaterium	no inhibition with sulfhydryl reagents	680527	-
additional information	Escherichia coli	no product inhibition by hydrazine	667115	-

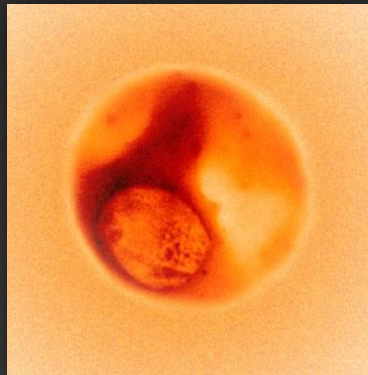
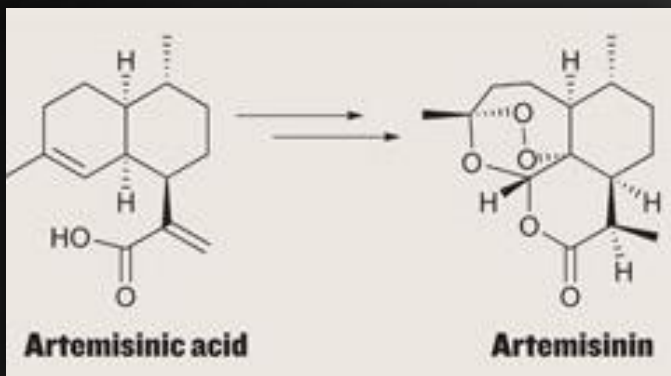
ACTIVATING COMPOUND	ORGANISM	COMMENTARY	LITERATURE	IMAGE
[bmim]dca	Escherichia coli	the enzyme activity in 25% [bmim]dca is 1.6fold than activity in water. The enzyme activity is decreased in higher concentration of [bmim]dca	685630	2D-image
additional information	Alcaligenes faecalis	addition of chemically synthesized fragments of the pro-peptide to purified mature enzyme increases its specific activity up to 2.3fold	651166	-

KM VALUE [mM]	KM VALUE [mM] Maximum	SUBSTRATE	ORGANISM	COMMENTARY	LITERATURE	IMAGE
2.5	-	2-benzoxazol-3-yl-acetic acid methyl ester	Escherichia coli	pH 6.8, 25°C	649527	2D-image
1.6	-	2-nitro-5-phenoxyacetamide benzoic acid	Streptomyces mobaraensis	-	687868	-
0.015	-	2-nitro-5-[(phenylacetyl)amino]-benzoic acid	Escherichia coli	pH 7.0, 30°C	668439	2D-image
0.00063	-	2-phenylacetamidobenzoic acid	Escherichia coli	25°C, pH 7.0, 10% dimethylsulfoxide	668459	2D-image
0.0004	-	3-phenylacetamidobenzoic acid	Escherichia coli	25°C, pH 7.0, 10%	668459	2D-image

Examples of metabolic engineering by synthetic biology

1. Yeast production of artemisinic acid for anti-malarials
Jay Keasling Group
2. Bacterial production of iso-butanol biopetrols
James Liao Group
3. Bacterial production of fatty acids biodiesels
Jay Keasling Group

Artemisinin – a valuable therapeutic compound



- Artemisinin is a natural plant product from the 'sweet wormwood' plant (*A.annua*)
- It is highly effective against multi drug-resistant *Plasmodium falciparum* malaria
- Releases free-radicals in blood cells to kill off the malarial parasite
- *P.falciparum* malaria is a major problem, particularly in Africa

A.annua grows naturally only in China and Vietnam

Harvest yields of artemisinin from *A.annua* are very low

High need but low supply and high cost

Difficult and expensive to do complete chemical synthesis

Synthesis from Artemisinic Acid pre-cursor is cheap and easy

Artemisinin acid – designing its biosynthesis

Can the difficult synthesis of artemisinin be done by microbes?

A. annua converts sugars into artemisinin

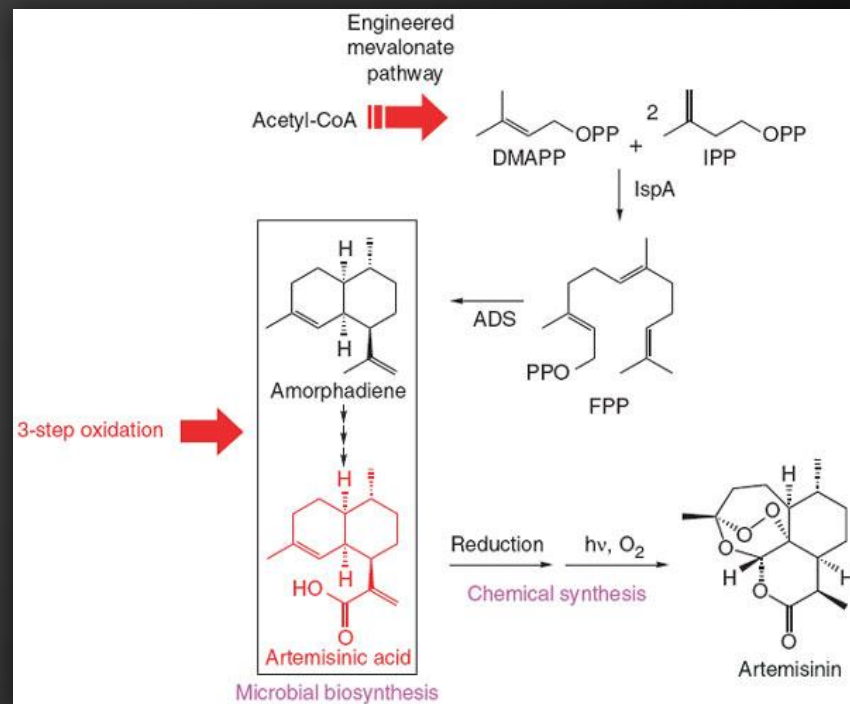
→ Import *A. annua* metabolic pathways into *E. coli* or Yeast

Artemisinin from artemisinic acid can be done easily by chemistry

Artemisinic acid comes from amorphaadiene
Amorphaadiene is made from isoprenoids

Isoprenoids are synthesised through the mevalonate pathway, already common to yeast and *E. coli*

Engineer native pathways and add *A. annua* enzymes to make artemisinic acid



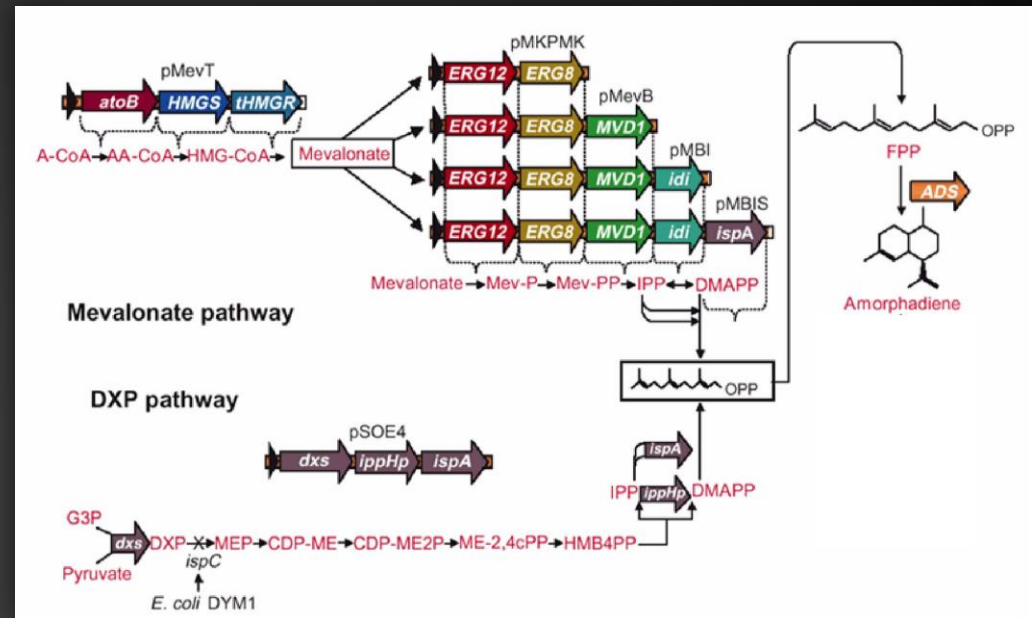
Re-engineering microbes to make amorphaadiene

Yeast and *E.coli* mevalonate pathway enzymes were engineered into operon units

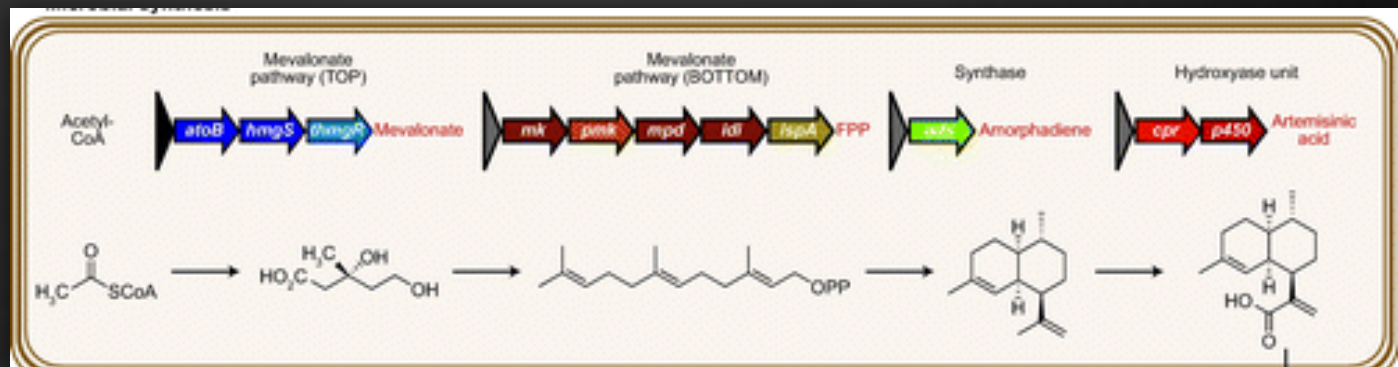
Each operon controlled by a promoter and on a plasmid

Heterologous pathways created alongside existing isoprenoid pathway

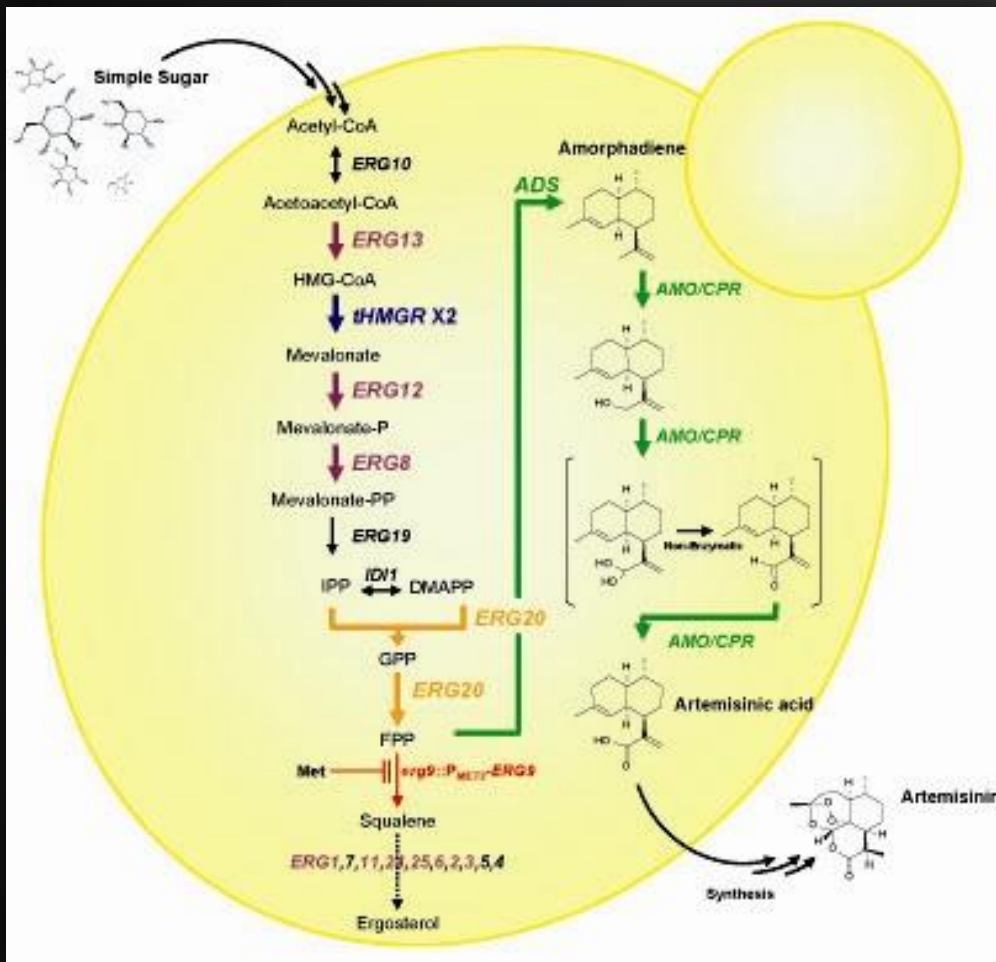
A.annua genes added to make product



Done in *E.coli* and Yeast



Semi-synthesis of artemisinin by engineered yeast



- Blue gene = directly up-regulated and dual-copy
- Purple gene = indirectly up-regulated by *upc2-1* – a global yeast biosynthesis regulator transcription factor
- Red line = strain has repression built-in
- Green = *A. annua* enzymes codon-optimized for yeast

DK Ro, EM Paradise et al., Nature 2006

Codon-optimisation

		Seconded Position									
		U		C		A		G			
		code	Amino Acid	code	Amino Acid	code	Amino Acid	code	Amino Acid		
First Position	U	UUU	phe	UCU	ser	UAU	tyr	UGU	cys	U	
		UUC		UCC		UAC		UGC		C	
		UUA	leu	UCA		UAA	STOP	UGA	STOP	A	
		UUG		UCG		UAG	STOP	UGG	trp	G	
	C	CUU	leu	CCU	pro	CAU	his	CGU	arg	U	
		CUC		CCC		CAC		CGC		C	
		CUA		CCA		CAA	gln	CGA		A	
		CUG		CCG		CAG		CGG		G	
	A	AUU	ile	ACU	thr	AAU	asn	AGU	ser	U	
		AUC		ACC		AAC		AGC		C	
		AUA		ACA		AAA	lys	AGA	arg	A	
		AUG	met	ACG		AAG		AGG		G	
	G	GUU	val	GCU	ala	GAU	asp	GGU	gly	U	
		GUC		GCC		GAC		GGC		C	
		GUA		GCA		GAA	glu	GGA		A	
		GUG		GCG		GAG		GGG		G	

Codon	Human	Drosophila	E. coli
Arginine:			
AGA	22 %	10 %	1 %
AGG	23 %	6 %	1 %
CGA	10 %	8 %	4 %
CGC	22 %	49 %	39 %
CGG	14 %	9 %	4 %
CGU	9 %	18 %	49 %
Total number of arginine codons	2403	506	149
Total number of genes	195	46	149

Efficiency of translation is affected by the codon used
 Coding sequence of a protein has an effect on how much is produced

Tuning the enzymes of the isoprenoid pathway

Key pathway to artemisinin – isoprenoids

How to up-regulate without creating bottlenecks and toxins?

Need to tune expression of each enzyme

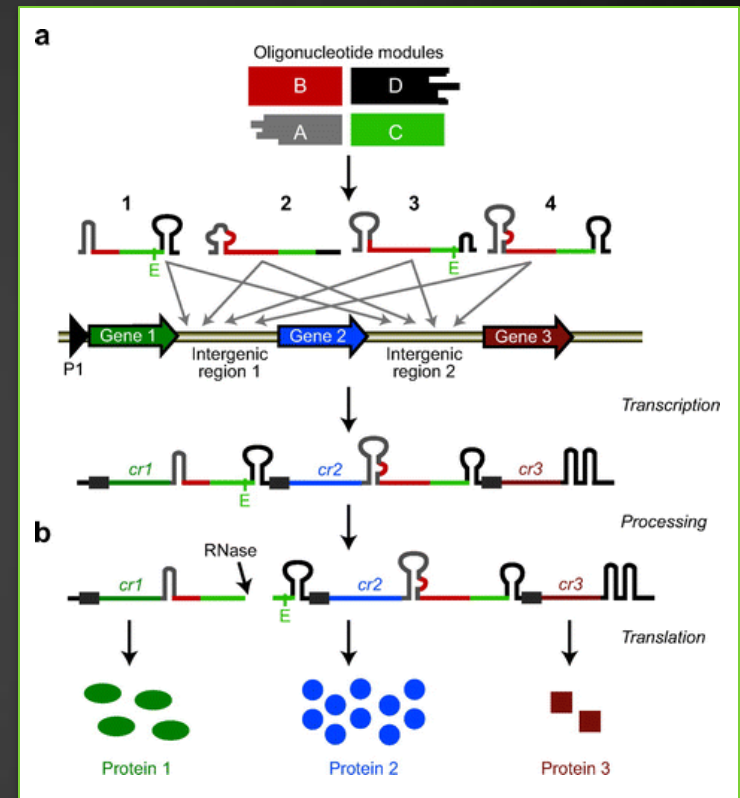
- Separate promoters for each?
- Directed evolution on each enzyme?

'Tunable Intergenic Regions' (TIGRs)

Modular RNA system to allow genes under the control of the same promoter to have different expression levels.

Modular = shuffling = directed evolution

Works in *E.coli* and yeast



BF Pfeiffer et al., Nature Biotech 2006

Beyond artemisininic acid – the value of isoprenoids

Gates Foundation, Sanofi-Aventis - Africa

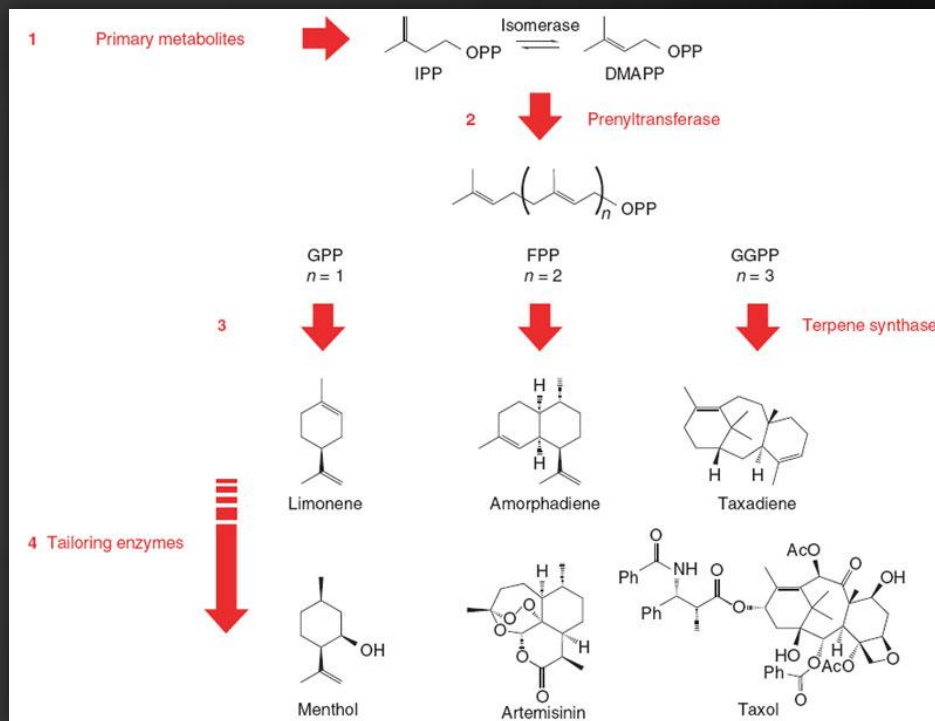
Full synthesis to Artemisinin

Improved synthesis:
every ATP counts

Other plant products – Taxol

Industry - Amyris

(biodiesels, aviation fuels and more)



Increased synthesis of mevalonate with scaffolds

Mevalonate synthesis in *E.coli*

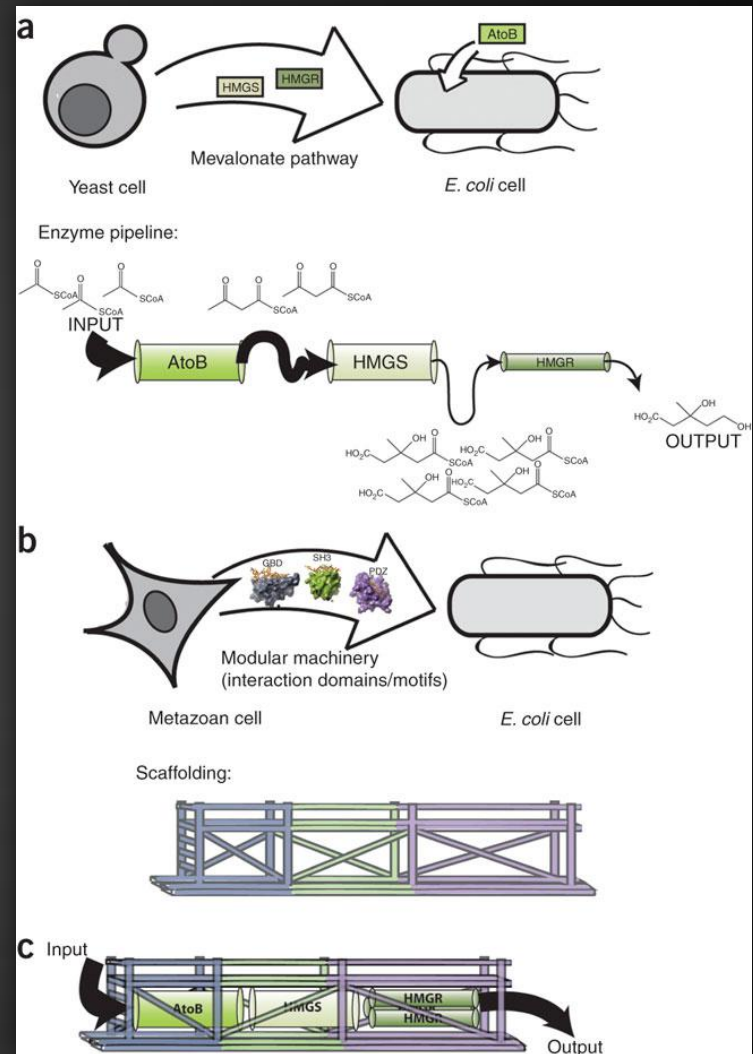
Create a 3-enzyme mevalonate pathway:
1 x *E.coli*, 2 x codon-optimized Yeast

Fuse each enzyme to a binding domain
Co-express a modular yeast scaffold protein

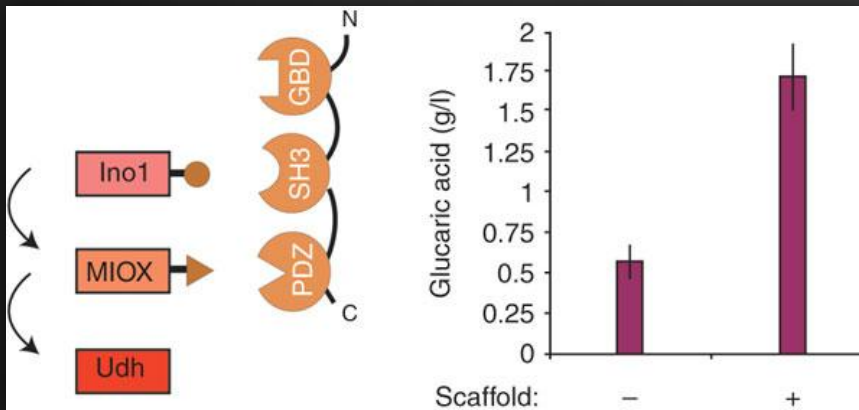
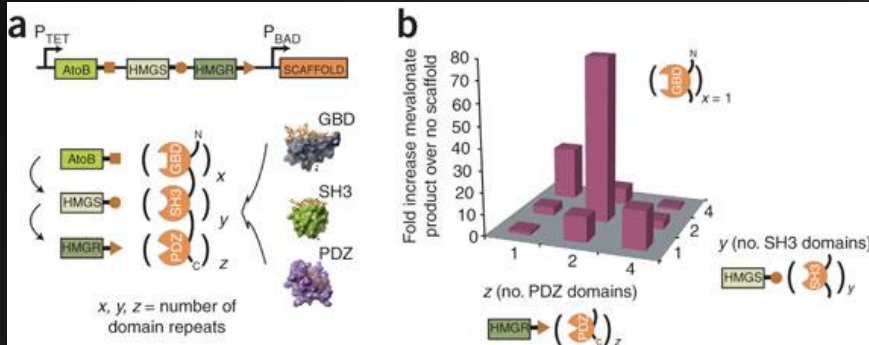
Scaffold protein binds all 3 enzymes next to each other

1. Increases pathway flux
2. Reduces toxic intermediate build-up
3. Prevents negative feedback inhibition from metabolite accumulation

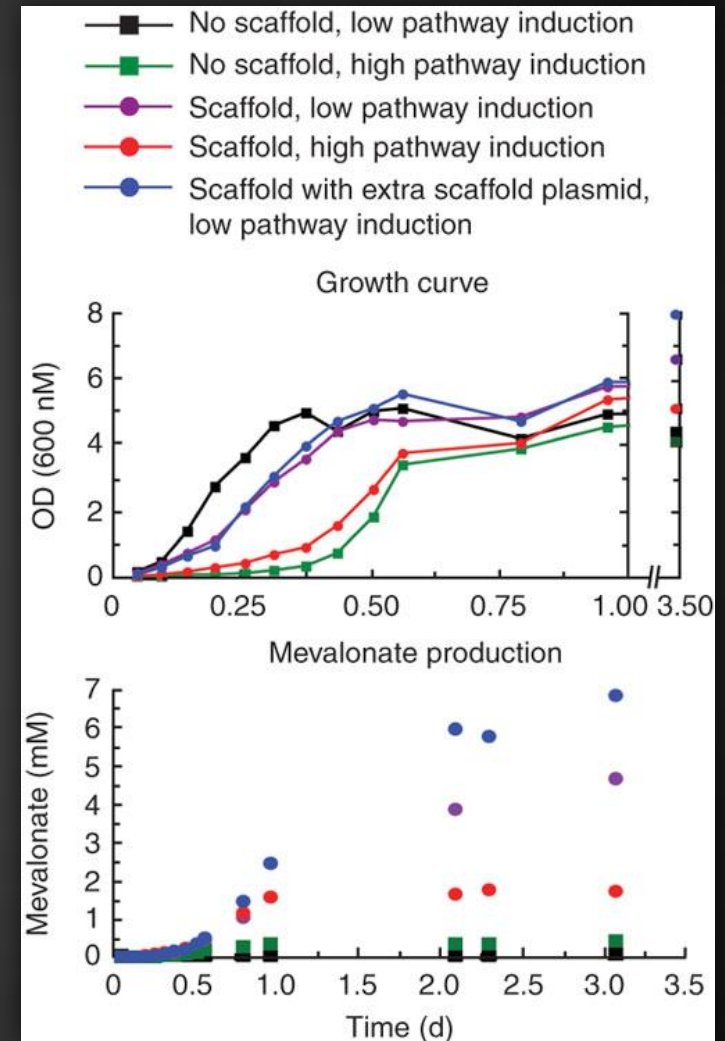
Dueber et al., Nature Biotech 2009



Mevalonate and glucaric acid synthesis with scaffolds



Dueber et al., Nature Biotech 2009

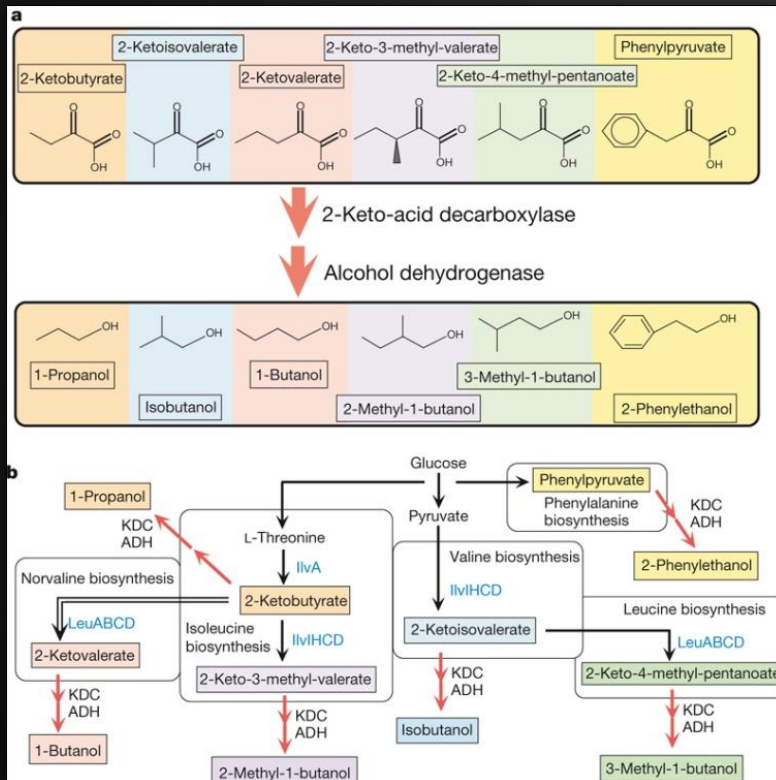


Petrol from *E.coli* – engineering iso-butanol production

Iso-butanol can be a direct replacement for petrol

Naturally synthesized by some strains of *Clostridium* – one of the oldest biotechnologies

Not produced naturally by any non-fermenting cells growing on glucose



- Use keto-acid pathways (amino acids synthesis)
- Overexpress entire pathway operons on plasmids
- Delete (knock-out) by-product forming enzymes

- Add a variety of different decarboxylases and dehydrogenases and look for product yields: *E.coli*, Yeast, *Clostridium*, *L.lactis*, *B.subtilis*

Table 1 | Alcohol production with KDC and ADH in *E. coli*

Product (μM)	KDC/plasmid				
	Kivd/pSA55	Aro10/pSA56	Pdc6/pSA49	Thi3/pSA57	Pdc (<i>C. acetobutylicum</i>)/pSA58
1-Propanol	520	290	125	ND	ND
Isobutanol	5,242	2,094	260	ND	75
1-Butanol	220	95	ND	ND	ND
2-Methyl-1-butanol	766	652	56	ND	ND
3-Methyl-1-butanol	1,495	1,099	92	ND	ND
2-Phenylethanol	324	469	ND	ND	175

The strain was JCL16 with various *kdc* genes and *S. cerevisiae* ADH2 expressed from plasmids. Culture was grown in M9 medium with 0.2 M glucose plus 0.1 mM IPTG at 30 °C for 40 h. These products were identified by GC-MS and quantified by GC-FID (see Methods). ND, not detectable.

Table 2 | Alcohol production with the supply of 2-keto acids

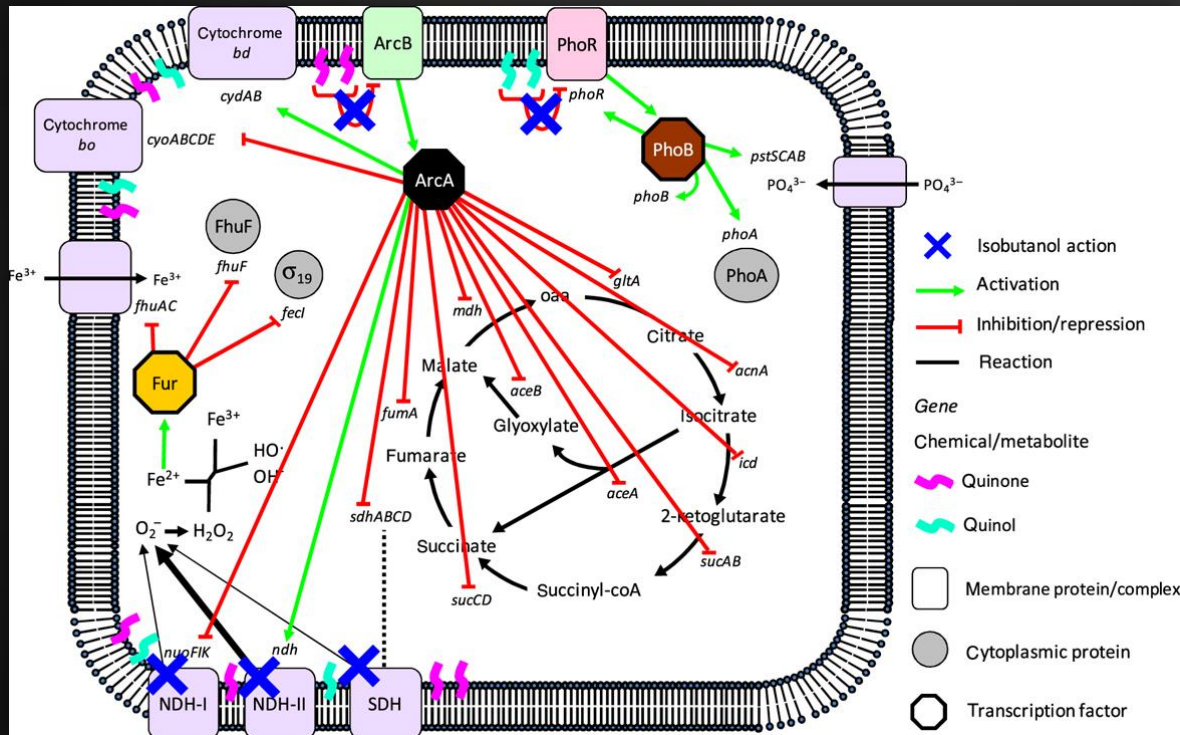
Product (μM)	2-Ketobutyrate	2-Keto-isovalerate	2-Ketovalerate	2-Keto-3-methyl-valerate	2-Keto-4-methyl-pentanoate	Phenylpyruvate
1-Propanol	2,138	ND	ND	ND	ND	8
Isobutanol	98	10,016	ND	ND	ND	64
1-Butanol	492	ND	3,926	ND	ND	23
2-Methyl-1-butanol	1,315	ND	ND	5,284	ND	ND
3-Methyl-1-butanol	ND	ND	52	ND	3,756	105
2-Phenylethanol	26	109	66	ND	ND	7,269

Strains and culture conditions are the same as described in Table 1. A total of 8 g l⁻¹ of 2-keto acids was added, except for 2-ketovalerate, where 1 g l⁻¹ was added because of its toxicity. ND, not detectable.

S Atsumi et al., Nature 2008

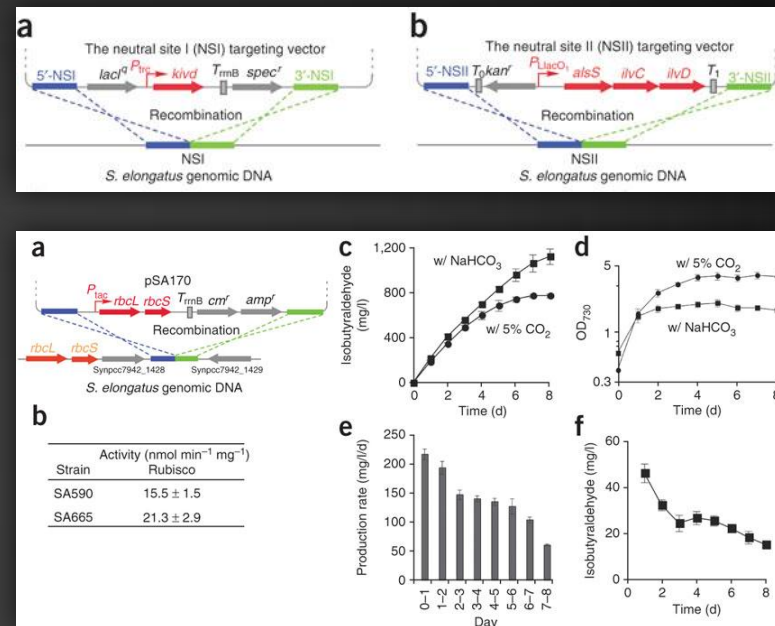
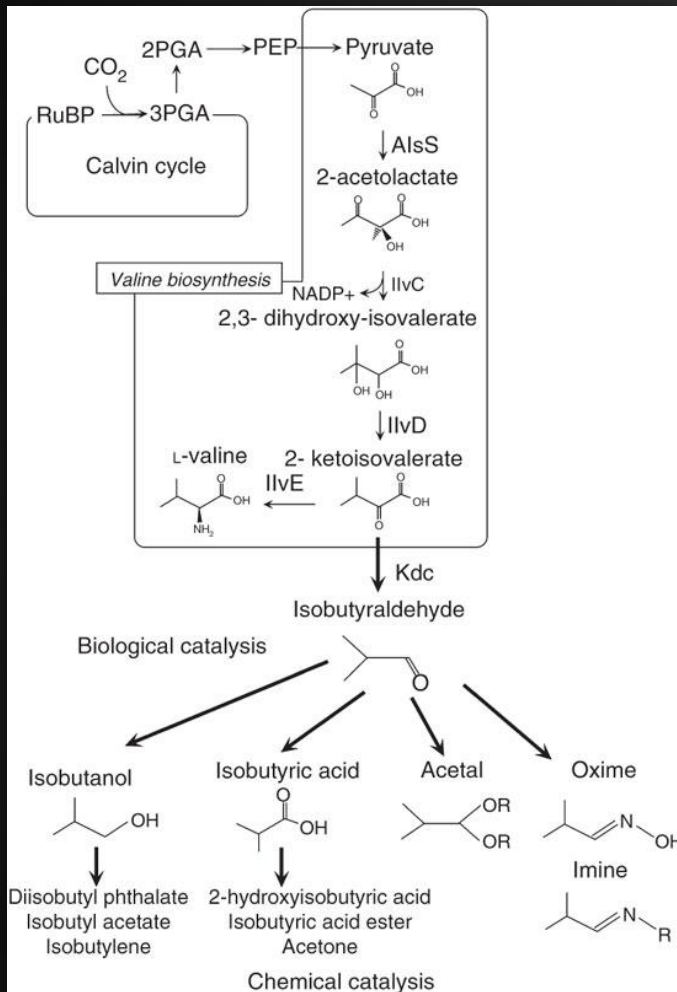
Petrol from *E.coli* keto-acids – systems biology analysis

- Network analysis follow-up study
- How does gene expression change when iso-butanol production is added?
- Do expression array studies and build a network map to indentify master controllers



MP Brynildsen and JC Liao, Mol Sys Biol 2009

Petrol from CO₂ – extending iso-butanol production



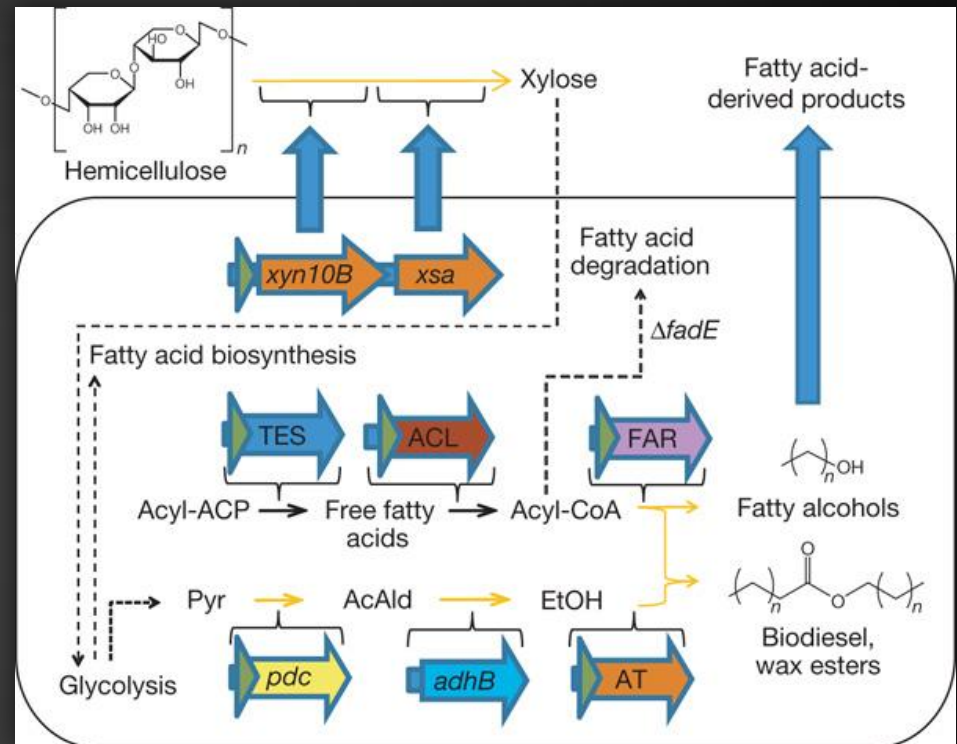
S Atsumi et al., Nature Biotech 2009

Synechococcus elongatus – cyanobacteria

- Overexpress related Rubisco enzymes
- Add keto-acid genes from *E.coli*, *B.subtilis*, *L.lactis*
- Chromosomal integration required

Fatty-acid biodiesel production in plant-digesting *E.coli*

1. Remove feedback inhibition from native fatty acid synthesis
 - Overexpress and free thioesterase
 - Overexpress 1st step of degradation
2. Prevent fatty acid degradation
 - Delete 2nd step of pathway from the cell genome (*fadE*)
3. Convert fatty acids into fatty alcohols
 - Express codon-optimized mouse FAR enzyme
4. Assemble new pathway to synthesize ethanol
 - Express *Zymomonas* *pdc* and *adhB*



EJ Steen and Y Kang et al., Nature 2010

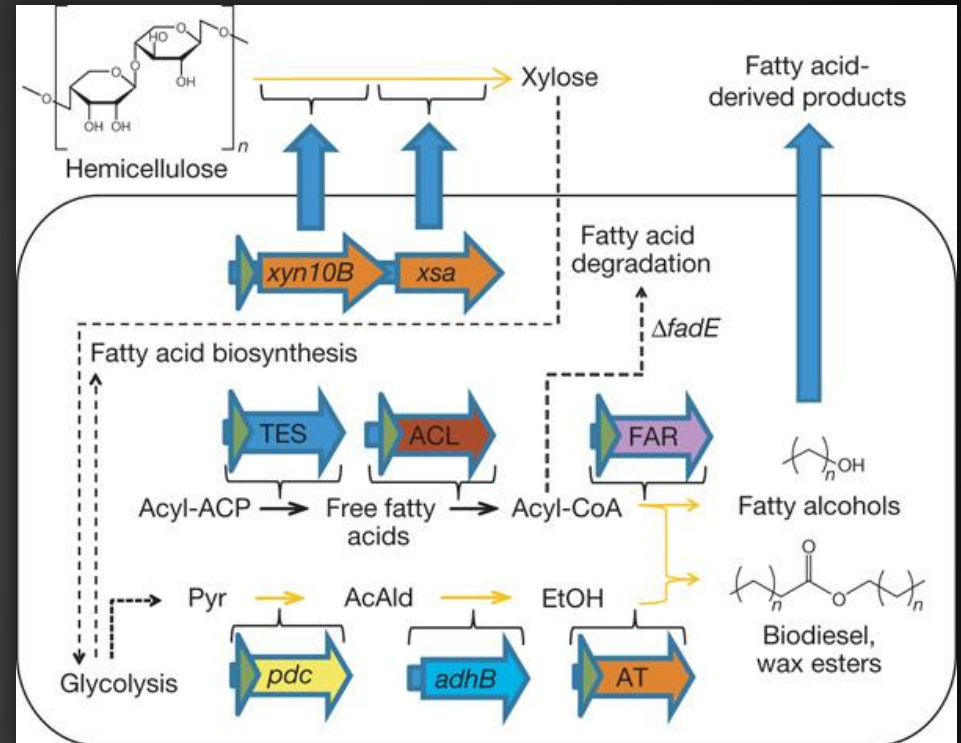
Fatty-acid biodiesel production in plant-digesting *E.coli*

- Combine ethanol and fatty acids and alcohols to get biodiesels, wax esters
 - Express *Acinetobacter* AT gene
- Use hemicellulose as a food source
 - Fuse codon-optimized cellulases from *Clostridium* and *Bacteroides* to cell surface proteins

Fatty products secrete from the cells

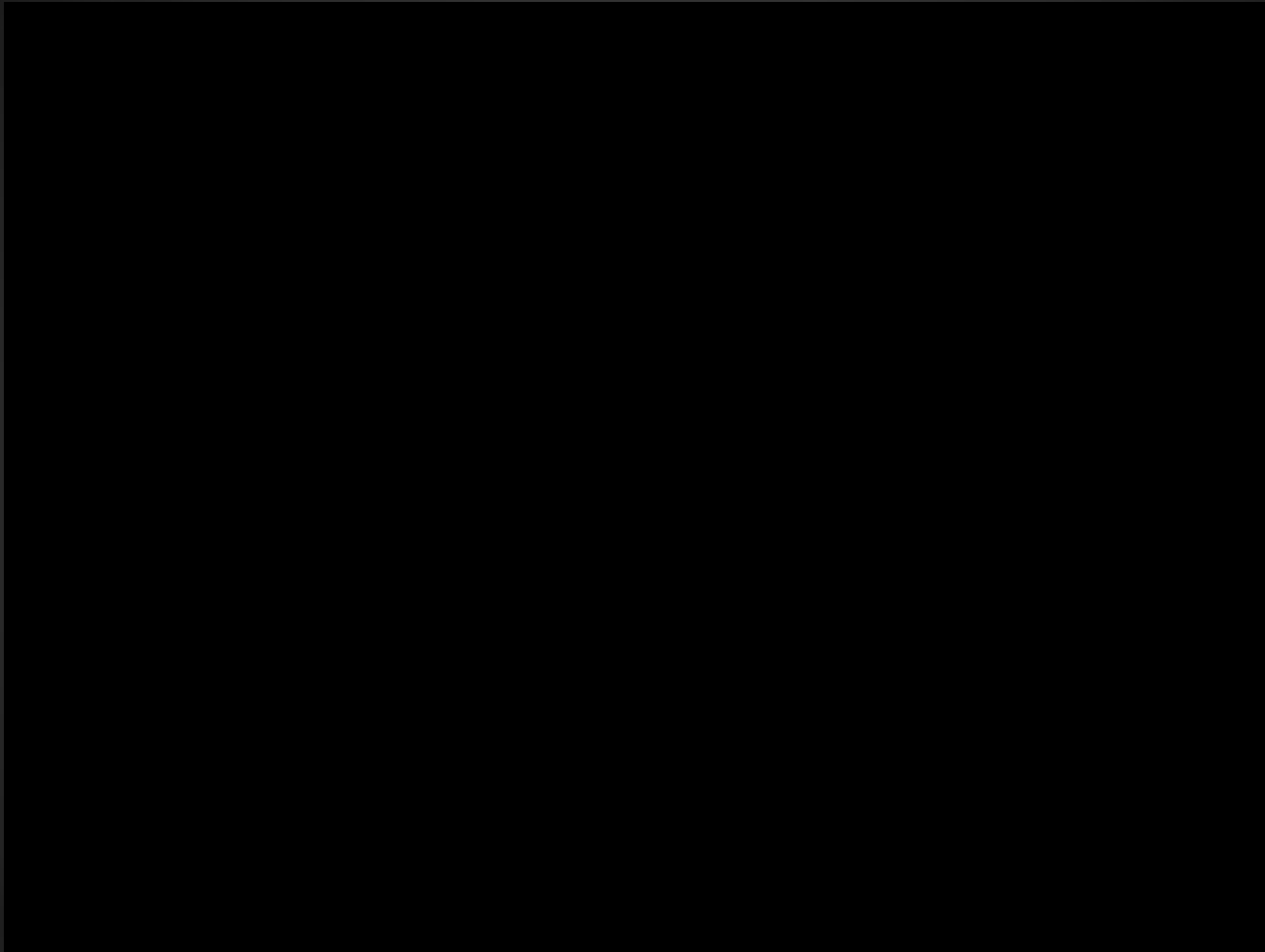
Swapping thioesterase (TES) from different organisms gives different fatty acid product lengths

All done by adding plasmids with genes with strong, regulated promoters



EJ Steen and Y Kang et al., Nature 2010

Fatty-acid biodiesel production in plant-digesting *E.coli*



BBC4 "The Cell" episode 3, 2009

Summary of synthetic biology used so far

- Up-regulate enzymes and pathways by cloning techniques
 - strong promoters, plasmids, codon-optimisation
- Add foreign enzymes from all across nature
 - create heterologous pathways, test out different versions of enzymes
- Co-ordinated expression of many enzymes
 - express genes from same regulated promoter
 - tune relative levels in an operon with TIGRs / IRES / regulation units
- Knock-out competing pathways and enzymes by modifying genomic DNA
- Use mutation/shuffling and selection to evolve increase yields
- Co-localize pathway enzymes on a scaffold to increase flux to product
- Combine FBA and systems biology to determine global regulators

The future for metabolic engineering

- The minimal cell – “tailor-made” cell chassis (JC Venter)
- Industrial metabolic engineering – e.g. LS9, Amyris
- Yeast – cytochromes, post-translation protein modifications
- Algae – biofuel from sunlight, hydrogen, even olive oil



- Automation of flux balance in pathways by shuffle/evolution
- Self-regulation built in to pathways
- Predictive network models linking genome to metabolic flux
- Consortia of engineered microbes growing with different roles

What you should now know and read up on!

You could get exam questions on...

1. The complexity and diversity of metabolism and how these are a challenge to engineering
2. The synthetic biology tools that can be used to overcome these challenges in metabolic engineering
3. The use of modeling, bioinformatics and systems biology in metabolic engineering
4. Examples of high-value products that could be made by metabolic engineering
5. How microbes have been engineered to help in the synthesis of artemisinin
6. Production of biofuels by engineered microbes (x2)

What you would synthesise, why and how?

Reading – useful reviews and perspectives

Toward engineering synthetic microbial metabolism – GH McArthur and SS Fong
Journal of biomedicine & biotechnology, Vol. 2010 (2010)

Synthetic Biology for Synthetic Chemistry – JD Keasling
ACS Chemical Biology, Vol. 3, No. 1. (1 January 2008), pp. 64-76.

Engineering for biofuels: exploiting innate microbial capacity or importing biosynthetic potential? – H Alper and G Stephanopoulos
Nature Reviews Microbiology, Vol. 7, No. 10. (01 October 2009), pp. 715-723.

Chemical synthesis using synthetic biology. – JM Carothers et al.
Current opinion in biotechnology, Vol. 20, No. 4. (31 August 2009), pp. 498-503.

Synthetic Metabolism: Engineering Biology at the Protein and Pathway Scales – CH Martin et al
Chemistry & Biology, Vol. 16, No. 3. (27 March 2009), pp. 277-286.

Advances in flux balance analysis – KJ Kauffman et al.
Current Opinion in Biotech, Vol. 14, No. 5. (October 2003), pp. 491-496.